Analysis of pre-service science teachers’ level of associating their chemistry knowledge with daily life and the relationship between these levels and their attitudes towards teaching science

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Abstract
The aim of this study is to determine science teacher candidates’ (PSTs) levels of associating their chemistry knowledge with daily life and to examine the relationship between these levels and their attitudes towards teaching science. The study is a descriptive study conducted through the survey method. This study was carried out with 30 PSTs in the part of science teacher education at the College of Education in their final semester. As the data collection tools, an open-ended questionnaire, which was developed by Kıyıcı and Aydoğdu (2011), was applied to determine the PSTs’ levels of associating their chemical knowledge with daily life. Also, the “Science Teaching Attitude Scale” developed by Thompson and Shringley (1986) and adapted into Turkish by Özkan, Tekkaya and Çakıroğlu (2002) was used to determine PSTs’ attitudes towards teaching science. The data obtained from the open-ended questionnaire analyzed based on the three categories that were used by Vazquez-Alanso and Manassero-Mas (1999). Findings obtained from the data showed that PSTs’ level of associating their chemistry knowledge with daily life was low. Results of the statistical analyses showed that there was a significant relations between level of associating their chemistry knowledge with daily and their attitudes towards teaching science.

Keywords: Science; teaching chemistry; chemistry of daily life; attitudes towards teaching science.

1. Introduction
The aim of the Science courses is to teach students scientific knowledge along with helping them to understand how to apply this knowledge to their daily lives (Pinarbaşı et all, 1998). People strive to understand and structure the world they live in throughout their lives. Studies on this situation showed that people interpret their world through learning processes that are conducted simply (Tversky & Kahneman, 1982; Todd & Gigerenzer, 2000; Leighton&Sternberg, 2004;

Stains & Talange, 2007). Associating science courses into daily life is also not a strong teaching strategy for teachers, but it has an outstanding contribution to significant and permanent learning (Mayoh & Knutton, 1997; Campbell & Lubben, 2000; McCann, 2001; Dogan, Kivrak & Baran, 2004). Chemistry subjects are crucial fields of science which are applied in all areas of life. The fact that students cannot fully understand chemistry, even though it has become expanded within our lives, has been stated by Huntemann et al. (1999) that the subjects of chemistry should be interpreted through daily life. Gilbert, Bulteb and Pilot (2011) underlined that a better understanding of chemistry subjects can be achieved by associating more frequently the subjects with daily life.

Science teachers have a crucial role in students’ understanding the subjects of chemistry and developing a positive attitude towards the chemistry. Emphasizing the role of the courses in students’ daily lives and associating the subjects with daily life during the lessons will contribute to significant and permanent learning. The aim of this study is to determine PSTs’ levels of associating their chemistry knowledge with daily life and to examine the relationship between these levels and their attitudes towards teaching science.

### 1. Aim of the Study

The aim of this study is to determine science teacher candidates’ level of associating the subjects of chemistry with daily life and to examine the relationship between these levels and their attitudes towards teaching science. The following research questions formed the basis for this study:

1) What is PSTs’ levels of associating their chemistry knowledge with daily life?
2) What is PSTs’ levels of associating their chemistry knowledge with daily life and the relationship between these levels and their attitudes towards teaching science?

### 2. Method

The screening method was selected to determine the extent to which pre-service science teachers can associate their chemistry knowledge with their daily lives. The screening method, which is among the non-empirical quantitative research methods, was used in this study (Johnson, 2001; Johnson and Onwueghuzie, 2004). The screening method is a research approach based on describing an event with its past or current state, comparing the relationship between the variables and collecting the data during a specific period of time (Karasar, 2002).

#### 2.1. Sample

The study sample consisted of 30 PSTs studying in their final year in the Part of Science Education Program participated in the study.

#### 2.2. Data Collection Tools:

A scale consisting of 20 open-ended questions, developed by Kıyıcı and Aydoğdu (2011) to determine PSTs’ levels of associating scientific information on physics, chemistry and biology with daily life, was used to determine PSTs’ levels of associating the subjects of chemistry with daily life in this study. After the required arrangements by consulting field experts, seven open-ended questions were selected and this questionnaire was completed.

The “Science Teaching Attitude Scale”, developed by Thompson and Shringley (1986) and adapted into Turkish by Özkan, Tekkaya, and Çakıroğlu (2002), was used to determine PSTs’ attitudes towards teaching science in the study. This scale was consisted of 21 items with five-point Likert-type scale ranging from “strongly agree” to “strongly disagree”.

3. Findings

The qualitative answers that PSTs gave to the open-ended questions analyzed based on the three categories (0, 1, 3.5 point) that were used by Vazquez-Alanso and Manassero-Mas (1999). 3.5 points were given for each scientific view, 1 point for each partial scientific view and 0 point for each non-scientific view.

<table>
<thead>
<tr>
<th>Scientific view</th>
<th>Partial scientific view</th>
<th>Non-scientific view</th>
</tr>
</thead>
<tbody>
<tr>
<td>Answer is scientifically sufficiently explained, with no partial understandings or misunderstandings.</td>
<td>Answer is scientifically partly explained, with misunderstandings.</td>
<td>Non-scientific, wrong expressions, misunderstandings. Or not answered.</td>
</tr>
<tr>
<td>3.5 points</td>
<td>1 point</td>
<td>0 points</td>
</tr>
</tbody>
</table>

The ranges of mean score that were considered in evaluating the findings of the science teaching attitudes scale were; “I strongly disagree” for 1.00-1.80, “I disagree” for 1.81-2.60, “I'm not sure” for 2.61-3.40, “I agree” for 3.41-4.20 and “I strongly agree” for 4.21-5.00.

The codes that indicate the opinions of PSTs about associating their scientific chemistry knowledge with daily life are given on Table 1.

Table 1: Score Distributions of PSTs’ Statements about Associating Chemistry Subjects to Daily Life

<table>
<thead>
<tr>
<th>Scientific view (3.5 points)</th>
<th>Partial view (1 point)</th>
<th>Non-scientific view (0 points)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hydrochloric acid is carried in plastic containers</td>
<td>18 60.00 8 26.67 4 13.33</td>
<td></td>
</tr>
<tr>
<td>We use plastic instead of glass containers to freeze water into ice.</td>
<td>23 76.67 6 30.00 1 3.33</td>
<td></td>
</tr>
<tr>
<td>Food is cooked faster in pressure cookers.</td>
<td>22 73.33 8 26.67 0 0.00</td>
<td></td>
</tr>
<tr>
<td>Soaps have slippery layers.</td>
<td>20 66.67 6 20.00 4 13.33</td>
<td></td>
</tr>
<tr>
<td>We place preserve jars in hot water when their lids won’t open.</td>
<td>22 73.33 8 26.67 0 0.00</td>
<td></td>
</tr>
<tr>
<td>Aluminum and steel are used in the production of saucepans and frying pans.</td>
<td>20 66.67 7 23.33 3 10.00</td>
<td></td>
</tr>
<tr>
<td>Icy roads are salted when it snows.</td>
<td>23 76.67 7 23.33 0 0.00</td>
<td></td>
</tr>
</tbody>
</table>

PSTs were asked to explain carrying hydrochloric acid with plastic containers according to the cause-effect relationship. 18 (60.00%) PSTs scientifically explained this question at a sufficient level. Eight (26.67%) PSTs gave a partly scientific answer and the remaining four (13.33%) PSTs were observed to give a wrong answer to the question. Examples of the answers that PSTs gave are as follows:

“...Hydrochloric acid is actually an acid chemical (HCl), this acid does not react with plastic objects and can be carried in plastic containers...” (PST23) (3.5 points)

“...Hydrochloric acid is sold in plastic containers, so it must be safe...” (PST20) (0 points)
In the study, the PSTs were asked, “We use plastic containers instead of glass containers when we want to freeze water into ice, explain this based on the cause-effect relationship”. 23 (76.67%) PSTs scientifically explained this question at a sufficient level. Seven (20.00%) PSTs gave partly scientific explanations for the question. The remaining one (3.33%) PST was observed to give a wrong answer. Examples of the answers that PSTs gave are as follows:

“...When water freezes, its density decreases and volume increases. The dilatation coefficient of plastic containers is higher than the dilatation coefficient of glass, so it expands more and doesn’t break like ice...” (PST16) (3.5 points)

“...I don’t know why glass breaks when water freezes but plastic containers don’t break...” (PST15) (1 point)

The PSTs were asked to explain why food is cooked faster in pressure cookers, based on the cause-effect relationship. 22 (73.33%) PSTs scientifically explained this question at a sufficient level. Eight (26.67%) PSTs gave partly scientific explanations for the question. Examples of the answers that PSTs gave are as follows:

“...Because steam doesn’t go out in pressure cookers the pressure inside increases and the boiling point of the liquid in the pressure cooker rises. Thus, foods are cooked faster...” (PST9) (3.5 points)

“...The steam in pressure cookers is kept inside that’s why it gets cooked faster...” (PST5) (1 point)

The PSTs were asked to explain why soap has a slippery layer. 20 (66.67%) PSTs scientifically explained this question at a sufficient level. Six (20.00%) PSTs gave partly scientific explanations for the question. The remaining four (13.33%) PSTs were observed to give a wrong answer. Examples of the answers that PSTs gave are as follows:

“...Soaps are basic substances. Basic substances give a slippery feeling to the hands. That’s why soaps have slippery layers...” (PST8) (3.5 points)

“...Soaps have a soft feature. That’s why they give a slippery feeling while on the hands...” (PST1) (0 points)

The PSTs were asked to explain why the lids of preserve jars can be opened after placing in hot water, based on the cause-effect relationship. 22 (73.33%) PSTs scientifically explained this question at a sufficient level. Eight (26.67%) PSTs gave partly scientific explanations for the question. Examples of the answers that PSTs gave are as follows:

“...The dilatation coefficients of glass and preserve jars are different. Lids expand more than glass while at the same temperature and that’s how the lids are able to be opened...” (PST16) (3.5 points)

“...When put in hot water, the lid slakes and opens...” (PST15) (1 point)

The PSTs were asked to give the reasons why substances such as aluminum and steel are used in the production of saucepans and frying pans. 20 (66.67%) PSTs scientifically explained this question at a sufficient level. Seven (23.33%) PSTs gave partly scientific explanations for the question. The remaining three (10.00%) PSTs were observed to give a wrong answer. Examples of the answers that PSTs gave are as follows:

“...Aluminum conducts heat regularly and it cools down fast. However, steel does not conduct heat very fast and is more durable against corrosion. I think it is preferred for these reasons...” (PST14) (3.5 points)

“...They must be preferred because they don’t cost much...” (PST2) (1 point)

The PSTs were asked to explain the reasons why icy roads are salted when it snows. 23 (76.67%) PSTs scientifically explained this question at a sufficient level. Seven (23.33%) PSTs gave partly scientific explanations for the question. Examples of the answers that PSTs gave are as follows:

“...The freezing point of salt is lower than the freezing point of ice. The melting point of the ice on road will be reduced when salted and this will enable the ice to melt...” (PST11) (3.5 points)
“...Salt is spread because it melts the ice on the road...” (PST5) (1 point)

It is evident from Table 2 that the attitudes of PSTs towards teaching science were 3.47. This value indicates that PSTs have a positive attitude towards teaching science.

### Table 2: PSTs’ Attitudes towards Teaching Science

<table>
<thead>
<tr>
<th>Attitudes Towards Teaching Science</th>
<th>N</th>
<th>X</th>
<th>S.d.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>30</td>
<td>3.47</td>
<td>.45</td>
</tr>
</tbody>
</table>

The Pearson correlation analysis was conducted to determine the relationship between PSTs’ level of associating their chemistry knowledge with daily life and their attitudes towards teaching science. According to Table 3, there is a positive significant relationship between PSTs’ levels of associating their chemistry knowledge with daily life and PSTs’ attitudes towards teaching science \((r=.711, p<.001)\).

### Table 3: The Relationships Between PSTs’ Attitudes Towards Teaching Science and Their Levels of Associating Chemistry Knowledge with Daily Life

<table>
<thead>
<tr>
<th>Attitudes Towards Teaching Science</th>
<th>Level of associating with daily life</th>
<th>.711**</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
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</tbody>
</table>

**. Correlation is significant at the 0.01 level (2-tailed).

### 4. Conclusions and Discussion

Students’ motivations towards their courses will increase when they understand what benefits their knowledge will bring to their daily lives (Parchmann et al, 2006). Thus, for students to understand the subjects, it is crucial that teachers teach scientific information by associating them with daily life (Kıyıcı&Aydoğdu, 2011; Thompson and Shrigley, 1986; Stavridou and Solomonidou, 1998; Ayas et al, 2001; Erdemir&Bakırcı, 2009). The extent to which learned knowledge is associated to daily life indicates how well the learning experience has taken place (Campbell&Lubben, 2000, Göçmençelebi İlkörücü&Özkan, 2009; Martin, 2009).

When the results of previous studies are considered, PSTs were observed to have failed to convey the scientific information about chemistry to daily life at sufficient levels. According to the literature, scientific knowledge is not fully conveyed to daily life (Pınarbaşi et al, 1998), Kişci&Aydoğan, 2011; Ayas& Özmen, 1998; Yiğit, Devecioğlu&Ayvacı, 2002). PSTs’ professional achievements depend on the extent to how well they teach the course subjects to their students. Associating the concepts about chemistry with daily life, rather than directly conveying them, will promote permanent learning and enable students to generate opinions about the different events they encounter in life (Göçmençelebi İlkörücü& Özkan, 2009; Coştu, Ünal&Ayas, 2007; Özmen, 2003; Sekerci&Canpolat, 2014). When the scientific knowledge that PSTs have acquired is not sufficiently conveyed to daily life, then they will not fully benefit their students throughout their professional experience.

Pearson correlation analysis was conducted to examine the relationship between PSTs’ level of associating their chemistry knowledge with daily life and their attitudes towards teaching science. It was observed that there is a positive significant relationship between conveying scientific knowledge to daily life and the attitudes towards teaching science. This result suggests that the attitude have an effect on the students’ success (Duit&Treagust, 2003; Bilgin&Aykac, 2016; Tuan, Chin&Shieh, 2005; Saricam&Sahin, 2015).
References


